

DoD IH Forum 2005: Estimating Service Life for Chemical Respirator Cartridges ...An Overview...

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Why do we need to estimate chemical cartridge service life ??

Respirator cartridges don't last forever...

Because OSHA says so !!

Per 29 CFR 1910.134 (d)(3)(iii)

Can no longer rely on warning properties...

 Does allow use of chemical cartridges against contaminant that don't have warning properties...



When do we change the cartridges ??

Two options for determining when to change respirator cartridges:

End of service life indicator

Change out schedule

End of Service Life Indicators

There are very few NIOSH-approved ESLI's:

Ammonia
Carbon monoxide
Ethylene oxide
Hydrogen chloride
Hydrogen fluoride
Hydrogen sulfide
Mercury
Sulfur dioxide
Toluene-2,4-diisocyanate
Vinyl chloride

Cartridge Change-out Schedules

Base the schedule on available objective info and data

Include in your written program:

 Data or information that was used to develop the schedule
 The basis for your use of that data

The best source of info is the manufacturer's test data

Cartridge Change-out: Methods for determining service life

Estimate with rules of thumb Test in laboratory using simulated workplace

Test cartridge in workplace

Test cartridge after use

Calculate using breakthrough equations (or use manufacturers' software)

Cartridge Change-out Rules of Thumb

Service life should be at least 8 hours if:
Contaminant has BP > 70 C (158 F)
Concentration is < 200 ppm
Relative humidity < 50%

Flow rate

 Service life is inversely proportional to flow

Cartridge Change-out Rules of Thumb

Concentration

 Reducing the concentration by a factor of 10, increases service life by a factor of 5

Humidity

 Humidity greater than 85% reduces service life by 50%

Experimental Testing in Lab or Field

 Most reliable & accurate, especially for mixtures

Can be used to validate existing change schedule

Is equipment, labor, & time intensive

Can be \$\$\$

Mathematical Model: Wood's Equation

- $t_b = breakthrough time (min)$
- $C_x = exit \text{ concentration } (g/cm^3)$
- $C_o = inlet concentration (g/cm^3)$
- Q = volumetric flow rate (cm³/min)
- W = weight of carbon adsorbent (g)
- ρ_{β} = bulk density of the packed bed (g/cm³)
- $W_e = equilibrium adsorption capacity (g/g carbon)$
- $k_v = adsorption rate coefficient (min^{-1}).$

SHA Service Life Calculator

Use on-line or download "Advisor Genius" program at www.osha.gov/SLTC/etools/respiratory/ advisor_genius_wood/ advisor_genius.html

Will need manufacturer's cartridge information

- Number of cartridges used by respirator
- Weight of sorbent in each cartridge (in grams)
- Carbon micropore volume (in cm³/gm)
- Density of packed bed (in gm/cm³)

Other information needed

- Max temperature & humidity expected in workplace
- Max concentration of contaminants in workplace (in ppm)
- Work-rate (volumetric flow rate or breathing rate in LPM)

Manufacturers' Service Life Calculators

NORTH Download "EZGuide" program at www.northsafety.com/usa/en/bs_sproduct. html?GID=3693

Email Scott Technical Support directly at techsupport.scotths.us@tycoint.com

Download "Merlin" program at

www.aosafety.com/aosafety.com/industrial/ resp_main.htm

Dealing with Mixtures...?

If breakthrough times for individual chemicals are within one order of magnitude (10X), add concentrations together and assume entire mixture behaves like the component with shortest individual breakthrough time.

TICIDA

 If breakthrough times for individual chemicals vary by two orders of magnitude (100X) or more, base service life on component with shortest individual breakthrough time.

OSHA Method for Mixtures

Employees are exposed to following:
 Chemical A = 100 ppm
 Chemical B = 100 ppm
 Chemical C = 75 ppm

Service life software predicts the following service times for each substance:
 Chemical A = 3770 minutes
 Chemical B = 3290 minutes
 Chemical C = 2480 minutes

OSHA Method for Mixtures

TICIDA

Breakthrough times are within one order of magnitude Add concentrations (= 275 ppm) Assume mixture behaves like component with shortest individual breakthrough time (Chemical C) Recalculate service life for Chemical C at 275 ppm = **989 minutes**

Mole Fraction Method for Mixtures

Total ppm of mixture = 275 ppm Mole fractions: Chemical A: 100 ppm ÷ 275 ppm = 0.36 Chemical B: 100 ppm ÷ 275 ppm = 0.36 Chemical C: 75 ppm ÷ 275 ppm = 0.27 • For each chemical, multiply the mole fraction by the single component breakthrough time...

Mole Fraction Method for Mixtures

Chemical	Mole Fraction	Single Substance Breakthrough Time	Breakthrough Time in Mixture
Chemical A	0.36	3770 minutes	1357 minutes
Chemical B	0.36	3290 minutes	1184 minutes
Chemical C	0.27	2480 minutes	670 minutes

Other Sources of Information

Navy Environmental Health Center at www-nehc.med.navy.mil/ih/ Respirator/Resp_index.htm

 OSHA Respiratory Protection eTool at www.osha.gov/SLTC/etools/ respiratory/ index.html

The End !!

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